

## Supplementary Materials

# Ionic liquid modified polymer gel for arsenic speciation

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## Figures

Section: 2.1. Synthesis and characterization of poly(MIA)

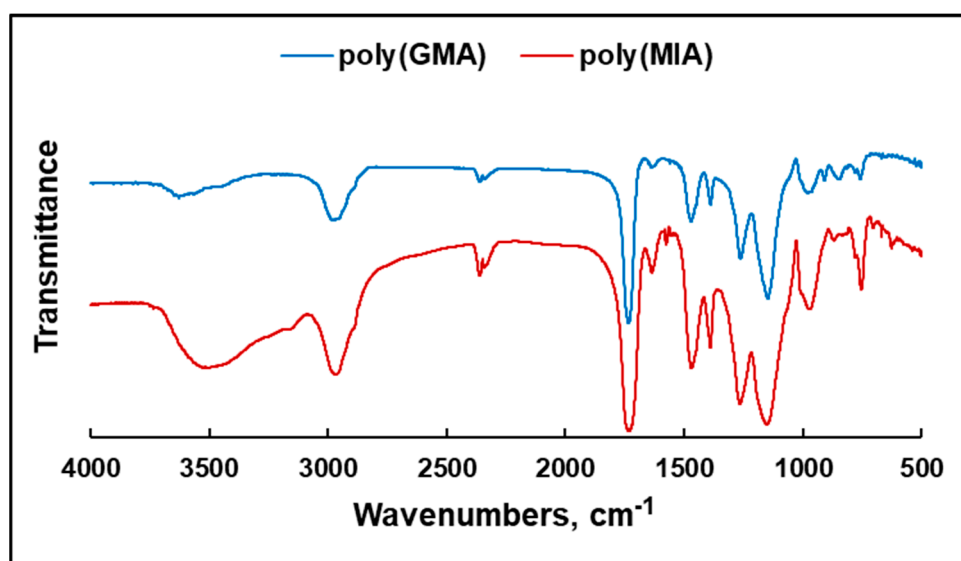
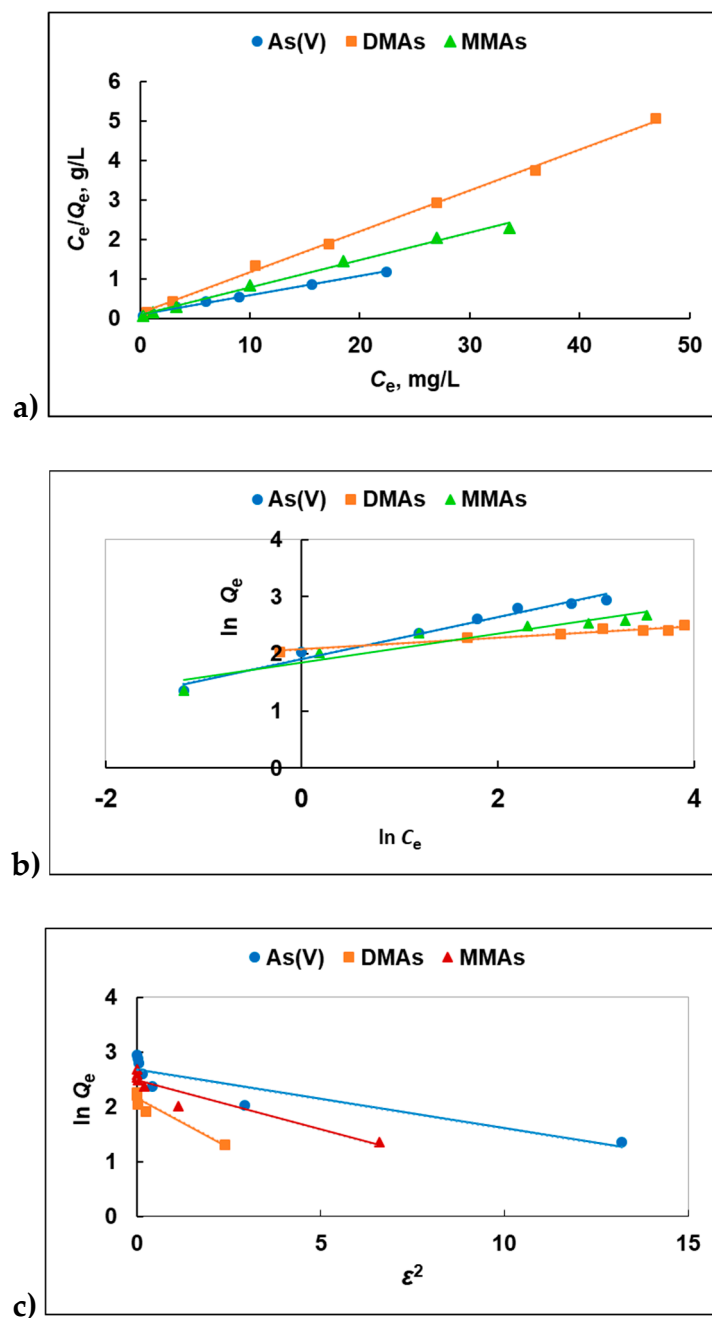


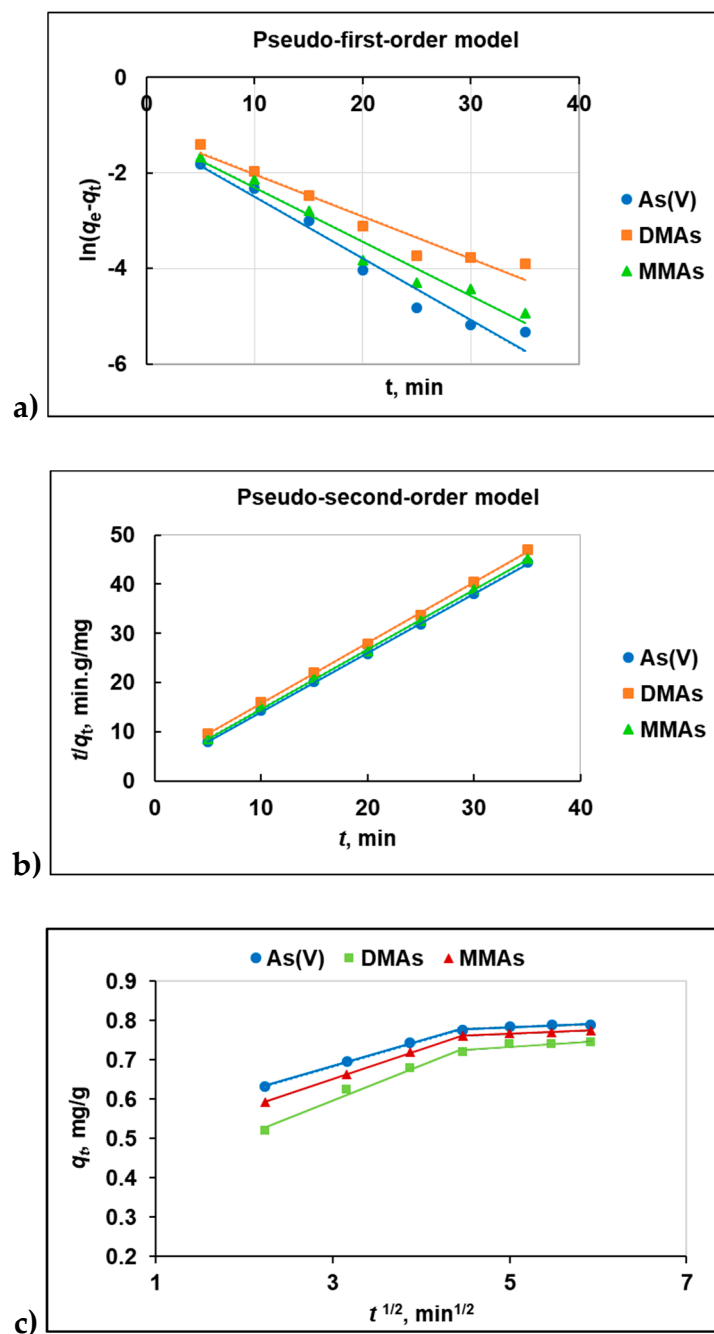
Figure S1. FTIR spectra of poly(GMA) and poly(MIA).

Section S2.4.1. Adsorption isotherm models



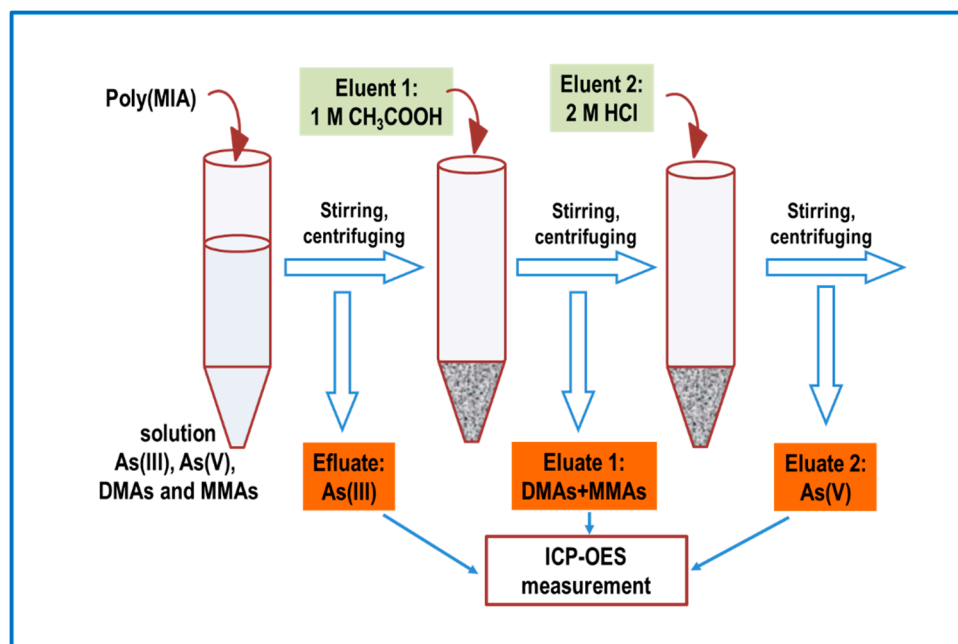
**Figure S2.** Langmuir (a), Freundlich (b) and Dubinin-Radushkevich (c) isotherms for adsorption of As(V), DMAs and MMAs on the poly(MIA).

Section S2.4.2. Modelling of As(V), DMAs and MMAs sorption kinetics



**Figure S3.** Adsorption kinetics of As(V), DMAs and MMAs ions onto the poly(MIA) at concentration 2 mg/L, pH 8, temperature 25 °C, and adsorbent dose 25 mg: (a) pseudo-first order; (b) pseudo-second order model; (c) intra-particle diffusion model.

**Section: 2.5. Analytical Application**



**Figure S4.** Scheme of the analytical procedure for As(III), As(V), DMAs and MMAs separation and determination in water samples using poly(MIA).

## Tables

### Section: 2.5. Analytical Application

**Table S1.** Added/Found method applied for water samples (three parallel determinations)

Sample	Added, µg/L			Found, µg/L		
	As(V)	As(III)	(DMA <sub>s</sub> +MMA <sub>s</sub> )	As(V)	As(III)	(DMA <sub>s</sub> +MMA <sub>s</sub> )
Lake Ognyanovo	1.0	-	0.1	1.1±0.1	<DL	0.09±0.01
	5.0	1.0	0.2	4.9±0.2	1.2±0.1	0.18±0.02
	0.5	0.2	0.1	0.4±0.1	0.18±0.02	0.11±0.02
River Iskar	1.0	-	0.1	0.9±0.1	<DL	0.12±0.02
	5.0	1.0	0.2	4.8±0.2	1.1±0.1	0.19±0.02
	0.5	0.2	0.1	0.5±0.1	0.21±0.02	0.09±0.02
Black sea water	2.0	0.1	0.05	1.9±0.2	0.11±0.01	0.044±0.003
	4.0	0.2	0.05	3.9±0.3	0.19±0.02	0.052±0.003
	10.0	0.5	0.1	10.1±0.9	0.51±0.03	0.09±0.01

**Table S2.** Recoveries achieved for arsenic species (model solutions) in the presence of environmentally relevant concentrations of anions in surface waters (three parallel experiments).

<b>Interferent</b>	<b>As(V)</b>	<b>MMA</b>	<b>DMA</b>
Cl <sup>-</sup> (500 mg/L)	>99	>99	>99
NO <sub>3</sub> <sup>-</sup> (50 mg/L)	98±2	96±2	97±2
HCO <sub>3</sub> <sup>-</sup> (300 mg/L)	96±4	94±3	95±3
SO <sub>4</sub> <sup>2-</sup> (200 mg/L)	92±4	93±4	95±4
PO <sub>4</sub> <sup>3-</sup> (20 mg/L)	93±5	92±4	93±4

**Section:** 2.6. Comparison with other methods

Table S3. Comparison of SPE methods using different sorbent materials for arsenic speciation.

Sorbent material	Separated species	pH	Sorbent mass (mg) / adsorbate concentration (mg/L)	Sorption capacity, mg/g	Analytical technique	Sample	LOD (µg/L)	Ref.
Alltech SAX and SCX columns	As(III)	6.5	500 / -	-	NAA	waters	0.9	[16]
	As(V)						1.7	
	DMA <sub>s</sub>						3.8	
	MMA <sub>s</sub>						1.6	
Three types immobilized macrocyclic materials	As(III)	7	- / 100 <sup>1</sup>	0.31-0.40 <sup>2</sup>	GF-AAS	waters	0.06	[17]
	As(V)						0.06	
	DMA <sub>s</sub>						0.05	
	MMA <sub>s</sub>						0.05	
Three types of resins: SBAE, HY-Fe and HY-AgCl	As(III)	7 – 9	1000 / 0.5	4.15	ICP-MS	waters	0.2	[18]
	As(V)			0.37			0.2	
	DMA <sub>s</sub>			0.10			0.2	
	MMA <sub>s</sub>			0.144			0.2	
A dual column packed with a magnetic MFC and MFC-SH	As(III)	6	- / 10 <sup>4</sup>	4.2	ICP-MS	Waters, hair, urine, SCC-7 cells	7.1 <sup>5</sup>	[19]
	As(V)			3.5			4.8 <sup>5</sup>	
	DMA <sub>s</sub>			2.4			6.3 <sup>5</sup>	
	MMA <sub>s</sub>			3.1			3.8 <sup>5</sup>	
tri-n-butylphosphine modified polymer microsphere	As(III)	7	- / 0.1 <sup>4</sup>	-	HPLC–ICP-MS	waters	1.2 <sup>5</sup>	[20]
	As(V)						0.91 <sup>5</sup>	
	DMA <sub>s</sub>						0.96 <sup>5</sup>	
	MMA <sub>s</sub>						0.82 <sup>5</sup>	
MGO-PSTH	As(III)	1	- / 5 <sup>4</sup>	1.6	HPLC-ICP-MS	Seawater	1.1 <sup>5</sup>	[21]

	As(V)			5.0			0.2 <sup>5</sup>	
	AsBet			1.1			3.8 <sup>5</sup>	
	DMA <sub>s</sub>			3.2			0.5 <sup>5</sup>	
NH <sub>2</sub> -UVM	As(III)	3.5	10 / 0.1 <sup>4</sup>	39.3 <sup>3</sup>	ETAAS	Waters,		[22]
	As(V)			123.2		human	3.3 <sup>5</sup>	
	TOA			44.2 <sup>3</sup>		urine		
MGO-IL	As(III)	9	20 / 50	160.65	AFS	water	-	[41]
	As(V)	2		104.13				
As(V) ion surface imprinted polymer	As(V)	3	20 / 10–800	39.5	UV-vis	waters	0.26	[42]
ionic liquid modified polymer gel	As(III)	8	25 / 2		ICP-MS	Waters	10 <sup>5</sup>	This
	As(V)			20.78			1 <sup>5</sup>	work
	DMA <sub>s</sub>			9.58			1 <sup>5,6</sup>	
	MMA <sub>s</sub>			14.50				

<sup>1</sup> μM; <sup>2</sup> mmol/g; <sup>3</sup> μg/g; <sup>4</sup> μg/L; <sup>5</sup> ng/L; <sup>6</sup> DMA<sub>s</sub>+ MMA<sub>s</sub>.

Abbreviations: AsBet - arsenobetaine; HY - hybrid resin; MFC - magnetic metal-organic framework composite; MFC-SH - mercapto-functionalized MFC nanoparticles; MGO-IL 1-butyl-3-methylimidazolium hexafluorophosphate ([BMIM] [PF<sub>6</sub>])-modified magnetic graphene oxide; MGO-PSTH - magnetic nanomaterial functionalized with [1,5-bis (2-pyridyl) 3-sulfophenylmethylene] thiocarbonohydrazide; NH<sub>2</sub>-UVM - amine-functionalized bimodal mesoporous silica nanoparticles; SAX - strongly anion exchanger; SCX - strongly cation exchanger; SBAE - strong base anion exchange resin; TOA - total organic arsenic (TOA (MMA, DMA, TMAO, AC and AB)